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$2 \text{Fe(s)} \rightarrow 2 \text{Fe}^{2+} + 4\text{e}^{-}$ $E^{\circ} = -0.44 \text{ V}$. Electrons released at anodic spot move through metal and go to another spot on the metal and reduce O_2 in presence of H^{+} . This spot behaves as cathode. $\text{O}_2(\text{g}) + 4 \text{H}^{+}(\text{aq}) + 4\text{e}^{-} \rightarrow 2 \text{H}_2\text{O(l)}$ $E^{\circ} = 1.23 \text{ V}$. The overall reaction is, $2 \text{Fe(s)} + \text{O}_2(\text{g}) + 4 \text{H}^{+}(\text{aq}) \rightarrow 2 \text{Fe}^{2+} + 2 \text{H}_2\text{O(l)}$ $E^{\circ} = 1,67 \text{ V}$

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Kerala Plus Two Chemistry Notes Chapter 2 Solutions. Solution A homogeneous mixture of two or more sub-stances is known as a solution. Solute : The substance present in smaller amount in a solution is called solute. Solvent: The substance present in larger amount in a solution is called solvent.

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Kerala Plus Two Chemistry Notes Chapter 9 Coordination Compounds. Coordination chemistry – branch of chemistry which deal with the complex compounds formed by transition and other metals. Chlorophyll, haemoglobin and vitamin B 12 are coordination compounds of Mg, Fe and Co respectively. Werner's Theory of Coordination Compounds The main postulates are,

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$2\text{Ca(OH)}_2 + 2\text{Cl}_2 \rightarrow \text{Ca(OCl)}_2 + \text{CaCl}_2 + 2\text{H}_2\text{O}$ Cl_2 is a powerful bleaching agent. $\text{Cl}_2 + \text{H}_2\text{O} \rightarrow 2\text{HCl} + [\text{O}]$ Coloured substance + $[\text{O}] \rightarrow$ colourless substance. Uses: For bleaching wood pulp, cotton and textiles; for the preparation of insecticide, pesticides and other organic solvents, e.g. CHCl_3 , DDT, BHC etc. Hydrogen Chloride (HCl): Preparation:

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In this thesis, the author introduces two strategies used to construct various types of N-heterocycles, based on the chemistry of zirconacycles and 2,6-diazasemibullvalenes. In the first part, the author presents the development of multi-component cyclization of a zirconacyclobutene-silacyclobutene fused compound, nitriles and

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unsaturated compounds. These reactions provide synthetically useful methodology for various N-heterocycles such as 3-acyl pyrrole, pyrrolo[3,2-d]pyridazine and dihydropyrroloazepine, which are all difficult to synthesize by other means. The isolation and characterization of the key three-fused-ring Zr/Si-containing intermediates are also described in detail. These results show that the zirconacyclobutene-silacyclobutene fused compound behaves as a “chemical transformer” upon treatment with various substrates via the “coordination-induced skeleton rearrangement” mechanism. In the second part, the author demonstrates the synthesis and isolation of a series of 2,6-diazasemibullvalenes (NSBVs) from the reaction of 1,4-dilithio-1,3-dienes and nitriles, highlighting the significant progress made for the first time in this work: (1) determination of X-ray crystal structure of a substituted 2,6-diazasemibullvalene; (2) measurement of the activation barrier of its rapid intramolecular aza-Cope rearrangement in solution; (3) exploration of several reaction types of NSBV with diverse ring-expansion products and “bowl-shape” or “cage-shape” N-containing polycyclic skeletons; (4) demonstration of the localized structure as the predominant form and the homoaromatic delocalized structure as a minor component in the equilibrium using theoretical analysis. Based on well-founded results, this work sheds new light on this controversial topic.

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The Chemistry of Ruthenium, Rhodium, Palladium, Osmium, Iridium and Platinum

The Chemistry of Lithium, Sodium, Potassium, Rubidium, Cesium, and Francium studies the physical and chemical properties of the elements listed in the title, including their chemical compounds and

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reactions. This book first features lithium, including its characterization, metals, and compounds. This topic is followed by discussions on the remaining featured elements in this text, encompassing their discovery and history, occurrence and distribution, and production. Then, this text presents the chemistry and chemical properties of the elements, specifically discussing topics such as the reactions of the metals, intermetallic compounds, hydrides, halides, cyanides and cyanates, and oxides and peroxides. The last two chapters examine biological activity and analytical chemistry of the elements. This book will be valuable to students and experts in the field of chemistry, as well as those in related fields.

Atmospheric Chemistry is a comprehensive treatment of atmospheric chemistry and covers topics ranging from the structure of the atmosphere to the chemistry of the upper atmosphere and the ionosphere. Atmospheric pollutants, hydrocarbon oxidation, and photochemical smog are also discussed, along with the reactions of O₃ and singlet O₂, the chemistry of SO₂ and aerosols, and methods for controlling atmospheric pollution. This book is comprised of 10 chapters and begins with an overview of the composition and chemistry of the atmosphere as well as its physical characteristics and the chemistry of meteors. The next two chapters deal with the chemistry of the upper atmosphere and the ionosphere, with emphasis on neutral oxygen atmosphere, carbon-hydrogen-oxygen cycle, and the D region. The chemistry of atmospheric pollutants is also examined, along with hydrocarbon oxidation and photochemical smog. The remaining chapters focus on the reactions of O₃ and singlet O₂, the chemistry of SO₂ and aerosols, and methods for controlling atmospheric pollution. This monograph

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should be useful to graduate students and scientists who wish to study atmospheric chemistry.

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