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ANSWER Answer

the following  
questions in the  
space provided.

1. The reaction

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for the  
decomposition of  
hydrogen

peroxide is  $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$ .

List three ways  
to speed up the  
rate of  
decomposition.

For each one,  
briefly explain  
why it is  
effective, based

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on collision  
theory.

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Questions.

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kinetic-

molecular theory

based on? ...

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relationship  
between  
atmospheric  
pressure and the  
actual boiling  
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Focus on this  
content, but  
make sure to  
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