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problem Basic
Thermodynamics-
Lecture

1_ Introduction
\u0026 Basic

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Thermodynamics
Chapter 1 – Lecture

1 Introduction and
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1) Understanding
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Thermodynamics I
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Physics #23

Thermodynamics,
PV Diagrams,
Internal Energy,
Heat, Work,
Isothermal,
Adiabatic, Isobaric,
Physics

Introduction to
Statics (Statics 1)
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Heat and Work -

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and Heat transfer

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associated with
thermodynamics
through the precise
definition of basic

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concepts to form a sound foundation for the development of the principles of thermodynamics.

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difference between

kg-mass and kg

force? Solution

Solution

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2 The science of
energy, that
concerned with the
ways in which
energy is stored

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within a body.

Energy transformations –
mostly involve heat
and work

movements. The
Fundamental law is
the conservation of
energy principle:
energy cannot be
created or
destroyed, but can
only be
transformed from

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another.
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observations

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whereas statistical
thermodynamics is

based on the

average behavior

of large groups of

particles. 1-2C On a

downhill road the

potential energy of

the bicyclist is

being converted to

kinetic energy,

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absolute pressure,
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manometer ...

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temperature of a system drops by 45°F during a cooling process.

Express this drop in temperature in K, R, and $^{\circ}\text{C}$.

Notes
March 15, 2016 -

Photons (light-waves) are emitted from an atom when an electron moves from a higher energy level to a

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lower energy level

o Energy = $h \times v$ -

Photons can also

be absorbed by an

atom when an

electron moves

from a lower

energy level to a

higher energy level

o Energy = $h \times v$ o

SAME ENERGY

LEVEL ...

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a system drops by
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