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Study Guide. You must show all work and setup for this to count as extra credit on  
your test (+3 points) 15.1 - 15.2 Heat, Calorimetry, and Enthalpy. Make the  
following conversions: 444 cal to joules. 1.8 kJ to joules. 0.45 kJ to calories.

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THERMOCHEMISTRY SECTION 17.1 THE FLOW OF ENERGY-HEAT AND WORK (pages 505-510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. ~ Energy Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that occur during chemical reactions? thermochemistry

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Chapter 17 Thermochemistry183 SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. Energy Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that

~~SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505–510)~~

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 $Q = m C \Delta T$   $\Delta H = -q C = 4184 \text{ J moles goC}$  1 When 150-g sample of KCl dissolves in 650 g of water in a calorimeter, the temperature drops from 310°C to 205°C  
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